

Asymmetric Hydrocyanation of Benzaldehydes Catalyzed by (5*R*)-5-(4-Imidazolylmethyl)-2,4-imidazolidinedione

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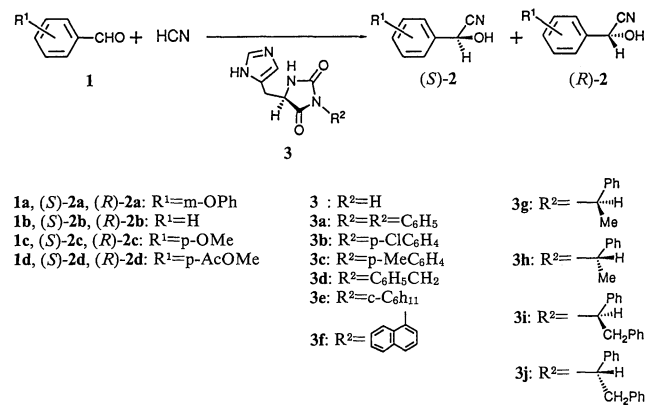
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Synopsis. The catalytic activity of (5*R*)-5-(4-imidazolylmethyl)-2,4-imidazolidinedione (**3**) was examined in the asymmetric hydrocyanation of 3-phenoxybenzaldehyde (**1a**) to (*S*)-2-hydroxy-2-(3-phenoxyphenyl)acetonitrile ((*S*)-**2a**), an important alcohol moiety of optically active pyrethroid insecticides. Among the catalysts, 3-benzyl derivative (**3d**) exhibited moderate enantioselectivities for **1a** and other benzaldehydes.

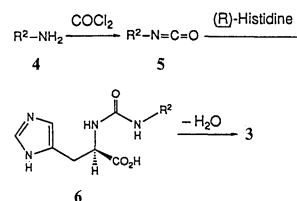
(*S*)-2-Hydroxy-2-(3-phenoxyphenyl)acetonitrile ((*S*)-**2a**) is an important alcohol moiety of optically active pyrethroid insecticides.¹⁾ With regard to the asymmetric synthesis of optically active cyanohydrins, there have been many reports concerning the enantioselective hydrocyanation or silylcyanation in the presence of a chiral catalyst.²⁾ Recently, Inoue et al. have reported that the chiral cyclic dipeptides containing an (*S*)-histidine residue, such as *cyclo*[(*S*)-phenylalanyl-(*S*)-histidyl], exhibit high enantioselectivities in the asymmetric hydrocyanation of aldehydes.³⁾ In conjunction with this finding, our interest has been focussed on the design and screening of catalysts containing the (*R*)-histidine residue to prepare (*S*)-**2a** in high optical yield. We thus prepared various kinds of (5*R*)-5-(4-imidazolylmethyl)-2,4-imidazolidinedione (**3**), which have a structural similar to that of *cyclo*[(*S*)-phenylalanyl-(*S*)-histidyl], and investigated their catalytic activities in the enantioselective addition of hydrogen cyanide to 3-phenoxybenzaldehyde (**1a**) as well as other benzaldehydes (**1b—d**) (Scheme 1). Herein, we report on the results concerning asymmetric hydrocyanation catalyzed by **3**.

Results and Discussion

The preparation of the catalyst (**3**) was carried out as illustrated in Scheme 2. The urea (**6**) was prepared by coupling of (*R*)-histidine monohydrochloride hydrate



Scheme 1.



Scheme 2.

with the isocyanate (**5**) which was obtained by a reaction of the corresponding amine (**4**) with phosgene. The urea (**6**) was dehydrated under acidic conditions to give **3** in reasonable yields. The asymmetric hydrocyanation was carried out by using 2.2—5 mol% of the catalyst (**3**) (based on the aldehyde (**1**)). The optical yield (ee) of (*S*)-**2** was determined by HPLC analysis (Sumipax OA-4100).⁴⁾

The asymmetric hydrocyanation of the aldehyde (**1a**) catalyzed by **3** was examined; the results are summarized in Table 1. In all of the catalysts, except for **3f**, (*S*)-cyanohydrin ((*S*)-**2a**) was preferentially obtained, while **3f** gave the opposite enantiomer ((*R*)-**2a**) as a major product (Entry 8). Among the catalysts (**3**), **3d** and **3e** gave (*S*)-**2a** in moderate optical yields (37% ee and 18% ee, Entries 4 and 6). Extending the reaction time decreased the optical yield, which may be due to a racemization of the products (Entry 5). The use of DMF as a solvent decreased the optical yield, even

Table 1. Asymmetric Hydrocyanation of **1a** Catalyzed by **3**^{a)}

Entry	Catalyst	Solvent	Time/h	Conv. of 1a / σ^b	Optical yield of 2a / σ^c ee (confign.) ^{c)}
1	3a	Neat	6	90	2 (<i>S</i>)
2	3b	Neat	4	88	1 (<i>S</i>)
3	3c	Neat	4	89	10 (<i>S</i>)
4	3d	Neat	1	90	37 (<i>S</i>)
5	3d	Neat	4	99	13 (<i>S</i>)
6	3e	Neat	4	22	18 (<i>S</i>)
7	3e	DMF	1	69	12 (<i>S</i>)
8	3f	DMF	4	91	6 (<i>S</i>)
9	3g	Neat	4	94	6 (<i>S</i>)
10	3h	Neat	4	97	2 (<i>S</i>)
11	3i	Neat	4	82	7 (<i>S</i>)
12	3j	Neat	4	73	1 (<i>S</i>)

a) The reactions were carried out at 10 °C by using 1.1 mmol of the catalyst (**3**), 99 mmol of hydrogen cyanide and 50 mmol of the aldehyde (**1a**). In Entries 7 and 8, 40 mL of DMF was used as a solvent because of a poor solubility of **3e** and **3f**. b) Determined by HPLC (LiChrosorb SI-60). Any by-product was not observed on HPLC. c) Determined by HPLC (Sumipax OA-4100).

Table 2. Asymmetric Hydrocyanation of **1a** Catalyzed by **3d**^{a)}

Entry	Solvent	Amount of 3d /mol vs. 1a	Temp/°C	Time/h	Conv. of 1a /%	Optical yield of (<i>S</i>)- 2a /%ee ^{c)}
1	Toluene	2.2	10	4	No reaction ^{d)}	—
2	Cyclohexane	2.2	10	4	No reaction ^{d)}	—
3	CH ₃ CN	2.2	10	1	89	17
4	CH ₂ Cl ₂	2.2	10	4	No reaction ^{d)}	—
5	DMF	2.2	10	1	92	11
6	Neat	2.2	0	3	91	40
7	Neat	5	0	1	90	41

a) The reactions were carried out by using 99 mmol of hydrogen cyanide and 50 mmol of **1a**. 40 mL of solvent was used in Entries 1—5. b) Determined by HPLC (LiChrosorb SI-60). Any by-product was not observed on HPLC. c) Determined by (Sumipax OA-4100). d) **3d** is insoluble.

Table 3. Asymmetric Hydrocyanation of Aldehydes Catalyzed by **3d**^{a)}

Entry	Aldehyde	Solvent	Time/h	Conv. of 1 /%	Optical yield of (<i>S</i>)- 2 /%ee ^{c)}
1	1b	Neat	1	90	33
2	1c	DMF	3	67	20
3	1d	DMF	1	93	16

a) The reactions were carried out at 0 °C by using 1.1 mmol of **3d**, 99 mmol of hydrogen cyanide and 50 mmol of aldehyde. 40 mL of DMF was used in Entries 2 and 3. b) Determined by HPLC (LiChrosorb SI-60). Any by-product was not observed on HPLC. c) Determined by HPLC (Sumipax OA-4100).

though the conversion of **1a** increased (compare Entry 7 with Entry 6). And even though the 3-substituent of **3d** was replaced by chiral benzyl groups, using **3g—j**, the optical yield was not improved (Entries 9—12).

Next, the optimal conditions were investigated in the asymmetric hydrocyanation of **1a** catalyzed by **3d**. The results are summarized in Table 2. No reaction was observed when toluene, cyclohexane or CH₂Cl₂ was used as a solvent, because of the poor solubility of **3d** in these solvents. The use of CH₃CN or DMF as a solvent decreased the optical yields. When the reaction temperature was lowered to 0 °C, the optical yield of (*S*)-**2a** was improved (40% ee, Entry 6), whereas it was not improved so much by increasing **3d** (41% ee, Entry 7).

The asymmetric hydrocyanation of benzaldehydes (**1b—d**) catalyzed by **3d** was examined. The results are exemplified in Table 3. In all cases, (*S*)-cyanohydrin ((*S*)-**2**) was obtained preferentially, ranging from 16% ee to 33% ee.⁴⁾ However, racemic cyanohydrin was obtained when isobutyraldehyde was employed for the reaction under the same reaction condition as Entry 1 (90% conv. at 1 h).

Thus, **3d** exhibited moderate enantioselectivities in the asymmetric hydrocyanation of benzaldehydes (**1**), particularly affording (*S*)-**2a** with a maximum optical yield of 41% ee.

We can discuss the mechanism of the enantioselective hydrocyanation catalyzed by **3d**, based on the mechanism which Inoue et al. postulated for the enantioselective hydrocyanation catalyzed by *cyclo*[(*S*)-phenylalanyl-(*S*)-histidyl].^{3b)} The carbonyl oxygen of

the aldehyde (**1**) is considered to coordinate to the catalyst (**3d**) by a hydrogen bond with the NH group of the histidine residue, and hydrogen cyanide interacts with the imidazolyl moiety of the histidine residue to form a cyanide ion which attacks the *re*-face of the activated carbonyl group, while the *si*-face is blocked by the aromatic ring of benzyl isocyanate residue. However, the blocking by the aromatic ring would be too loose to afford a satisfying enantioselectivity.

A further study dedicated to improving the enantioselectivity is in progress.

Experimental

General. The melting points are uncorrected. Optical rotations were taken on a JASCO DIP-140 digital polarimeter. ¹H NMR spectra were obtained at 200 MHz on a Varian XL-200, except for those of **3**, which were obtained at 90 MHz on a Hitachi R-40; ¹³C NMR spectra were obtained at 50.3 MHz on a Varian XL-200. IR spectra were obtained on a Hitachi 260-10. HPLC were recorded on a Shimadzu LC-6A, LiChrosorb SI-60 and Sumipax OA-4100 being used as columns for determination of the conversion of **1** and the optical yield of the resulting cyanohydrins ((*S*)-**2**: (*R*)-**2**), respectively. Toluene and xylene were distilled from sodium benzophenone ketyl. DMF and CH₃CN were distilled immediately prior to use. Hydrogen cyanide was provided by Ehime Factory. All other solvents and chemicals were used without further purification.

Preparation of (5*R*)-5-(4-imidazolylmethyl)-2,4-imidazolidinedione (3**).** (*5R*)-3-Benzyl-5-(4-imidazolylmethyl)-2,4-imidazolidinedione (**3d**). To a solution of 7.49 g (70 mmol) of benzylamine in 50 mL of xylene was added at 120 °C over 100 min 27.7 g (280 mmol) of phosgene. After removing the solvent and excess phosgene in vacuo, the crude isocyanate was subjected to the following procedure.

To a solution of 15.7 g (75 mmol) of (*R*)-histidine monohydrochloride hydrate in 180 mL of water was added 2 equiv NaOH until the solution showed pH 9. After the addition of the crude isocyanate obtained above, the reaction mixture was kept at 40 °C for 1 h, adjusted to pH 1 with 35% aqueous HCl, and kept under reflux for 1 h. After being cooled to 5 °C and neutralized to pH 5 with 2 equiv NaOH, the reaction mixture was allowed to stand at 5 °C overnight. The precipitate was filtered, washed with cold water and recrystallized from aqueous ethanol to give 15.2 g (85%) of **3d** as white crystals: Mp 219 °C; [α]_D²⁵ 10.6° (c 0.87, DMSO); IR (KBr) 3300, 1760, 1710, 1455, 1430 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ =2.95 (d, *J*=5.2 Hz, 2H), 4.36 (t, *J*=5.2 Hz, 1H), 4.47 (s, 2H), 6.80 (s, 1H), 6.9—7.4 (m, 5H), 7.52 (s, 1H), 8.17 (br.s, 1H). Found: C, 65.41; H,

5.49; N, 16.23%. Calcd for $C_{14}H_{14}N_3O_2$: C, 65.61; H, 5.51; N, 16.40%.

(5R)-5-(4-Imidazolylmethyl)-3-phenyl-2,4-imidazolidinone (3a). The procedure given for **3d** was carried out on the same scale using aniline to give 14.0 g (83%) of **3a** as white crystals: Mp 192–194 °C; $[\alpha]_D^{25}$ 151.1° (c 1.02, methanol); IR (KBr) 3200, 1760, 1700, 1430, 1190 cm^{-1} ; 1H NMR (DMSO- d_6) δ =3.08 (d, J =6.0 Hz, 2H), 4.44 (t, J =6.0 Hz, 1H), 6.90 (s, 1H), 7.2–7.5 (m, 5H), 7.60 (s, 1H), 8.20 (br.s, 1H). Found: C, 64.39; H, 4.87; N, 17.22%. Calcd for $C_{13}H_{12}N_3O_2$: C, 64.45; H, 4.99; N, 17.35%.

(5R)-3-(4-Chlorophenyl)-5-(4-imidazolylmethyl)-2,4-imidazolidinone (3b). The procedure given for **3d** was carried out on the same scale using 4-chloroaniline to give 14.3 g (74%) of **3b** as white crystals: Mp 220–221 °C; $[\alpha]_D^{25}$ 14.5° (c 0.86, DMSO); IR (KBr) 3300, 1710, 1660, 1570, 1400, 1190 cm^{-1} ; 1H NMR (DMSO- d_6) δ =3.01 (d, J =5.8 Hz, 2H), 4.45 (t, J =5.8 Hz, 1H), 6.89 (s, 1H), 7.1–7.7 (m, 4H), 7.73 (s, 1H), 9.00 (br.s, 1H). Found: C, 56.27; H, 3.89; N, 14.93%. Calcd for $C_{13}H_{11}N_3O_2Cl$: C, 56.43; H, 4.01; N, 15.19%.

(5R)-5-(4-Imidazolylmethyl)-3-(4-methylphenyl)-2,4-imidazolidinone (3c). The procedure given for **3d** was carried out on the same scale using *p*-toluidine to give 13.8 g (77%) of **3c** as white crystals: Mp 243 °C; $[\alpha]_D^{25}$ 113.7° (c 0.872, DMSO); IR (KBr) 3200, 3000, 1760, 1705, 1600, 1520, 1430, 1190 cm^{-1} ; 1H NMR (DMSO- d_6) δ =2.30 (s, 3H), 3.02 (d, J =5.8 Hz, 2H), 4.42 (t, J =5.8 Hz, 1H), 6.86 (s, 1H), 7.0–7.4 (m, 4H), 7.56 (s, 1H), 8.27 (br.s, 1H). Found: C, 65.59; H, 5.35; N, 16.17%. Calcd for $C_{14}H_{14}N_3O_2$: C, 65.61; H, 5.51; N, 16.40%.

(5R)-3-Cyclohexyl-5-(4-imidazolylmethyl)-2,4-imidazolidinone (3e). The procedure given for **3d** was carried out on the same scale using cyclohexylamine to give 13.9 g (80%) of **3e** as white crystals: Mp 172–174 °C; $[\alpha]_D^{25}$ 15.6° (c 0.87, H_2O); IR (KBr) 3300, 2950, 1620, 1580, 1400 cm^{-1} ; 1H NMR (DMSO- d_6) δ =0.8–0.9 (m, 11H), 2.90 (d, J =6.0 Hz, 2H), 4.33 (t, J =6.0 Hz, 1H), 6.93 (s, 1H), 7.90 (s, 1H). Found: C, 62.71; H, 7.19; N, 16.84%. Calcd for $C_{13}H_{18}N_3O_2$: C, 62.88; H, 7.31; N, 16.92%.

(5R)-5-(4-Imidazolylmethyl)-3-(1-naphthyl)-2,4-imidazolidinone (3f). The procedure given for **3d** was carried out on the same scale using 1-aminonaphthalene to give 16.6 (81%) of **3f** as white crystals: Mp 205–206 °C; $[\alpha]_D^{25}$ –39.4° (c 0.76, DMSO); 1H NMR (DMSO- d_6) δ =3.07 (d, J =6.2 Hz, 2H), 4.53 (t, J =6.2 Hz, 1H), 6.97 (s, 1H), 7.3–7.7 (m, 4H), 7.8–8.3 (m, 4H), 8.80 (br.s, 1H). Found: C, 69.76; H, 4.64; N, 14.25%. Calcd for $C_{17}H_{14}N_3O_2$: C, 69.85; H, 4.83; N, 14.38%.

(5R)-5-(4-Imidazolylmethyl)-3-[(R)-1-phenylethyl]-2,4-imidazolidinone (3g). The procedure given for **3d** was carried out on the same scale using (*R*)- α -methylbenzylamine to give 12.3 g (65%) of **3g** as white crystals: $[\alpha]_D^{25}$ –20.1° (c 1.00, DMSO); IR (KBr) 3300, 1755, 1630, 1550 cm^{-1} ; 1H NMR (DMSO- d_6) δ =1.33 (d, J =6.5 Hz, 3H), 2.88 (d, J =6.0 Hz, 2H), 4.36 (t, J =6.0 Hz, 1H), 4.73 (q, J =6.5 Hz, 1H), 6.82 (s, 1H), 7.15–7.4 (m, 5H), 7.65 (s, 1H). Found: C, 66.41; H, 5.80; N, 15.39%. Calcd for $C_{15}H_{16}N_3O_2$: C, 66.65; H, 5.97; N, 15.55%.

(5R)-5-(4-Imidazolylmethyl)-3-[(S)-1-phenylethyl]-2,4-imidazolidinone (3h). The procedure given for **3d** was carried out on the same scale using (*S*)- α -methylbenzylamine to give 11.5 g (61%) of **3h** as white crystals: $[\alpha]_D^{25}$ –36.8° (c 0.40, DMSO); IR (KBr) 3300, 1760, 1630, 1560 cm^{-1} ; 1H NMR (DMSO- d_6) δ =1.30 (d, J =6.5 Hz, 3H), 2.88 (d, J =6.0 Hz, 2H), 4.30 (t, J =6.0 Hz, 1H), 4.71 (q, J =6.5 Hz, 1H), 6.83 (s, 1H), 7.13–7.43 (m, 5H), 7.65 (s, 1H). Found: C, 66.60; H, 5.92; N, 15.47%. Calcd for $C_{15}H_{16}N_3O_2$: C, 66.65; H, 5.97; N, 15.55%.

(5R)-5-(4-Imidazolylmethyl)-3-[(R)-1,2-diphenylethyl]-2,4-imidazolidinone (3i). The procedure given for **3d** was carried out on the same scale using (*R*)-1,2-diphenylethylamine to give 13.2 g (55%) of **3i** as white crystals: $[\alpha]_D^{25}$ –15.1° (c 1.01, DMSO); IR (KBr) 3340, 1630, 1550, 1510, 1260 cm^{-1} ; 1H NMR (DMSO- d_6) δ =2.85 (d, J =6.5 Hz, 2H), 2.88 (d, J =6.5 Hz, 2H), 4.27 (t, J =6.0 Hz, 1H), 4.80 (t, J =6.5 Hz, 1H), 6.80 (s, 1H), 6.9–7.5 (m, 10H), 7.68 (s, 1H). Found: C, 72.68; H, 5.70; N, 11.99%. Calcd for $C_{21}H_{20}N_3O_2$: C, 72.81; H, 5.82; N, 12.13%.

(5R)-5-(4-Imidazolylmethyl)-3-[(S)-1,2-diphenylethyl]-2,4-imidazolidinone (3j). The procedure given for **3d** was carried out on the same scale using (*S*)-1,2-diphenylethylamine to give 12.9 g (53%) of **3j** as white crystals: $[\alpha]_D^{25}$ –37.0° (c 1.01, DMSO); IR (KBr) 3350, 1630, 1560, 1520, 1260 cm^{-1} ; 1H NMR (DMSO- d_6) δ =2.83 (d, J =6.5 Hz, 2H), 2.90 (d, J =6.0 Hz, 2H), 4.25 (t, J =6.0 Hz, 1H), 4.82 (t, J =6.5 Hz, 1H), 6.80 (s, 1H), 7.0–7.6 (m, 10H), 7.68 (s, 1H). Found: C, 72.77; H, 5.71; N, 11.95%. Calcd for $C_{21}H_{20}N_3O_2$: C, 72.81; H, 5.82; N, 12.13%.

General Procedure for Asymmetric Addition of Hydrogen Cyanide to Benzaldehyde (1) Catalyzed by 3. To a mixture of 1.1 mmol of the catalyst (**3**) and 50 mmol of the aldehyde (**1**) (if necessary, 40 mL of an appropriate solvent was used) at 5 °C was added dropwise 3.8 mL (2.67 g, 99 mmol) of hydrogen cyanide under nitrogen; stirring was continued at 0 °C (or 10 °C) for an appropriate period. The reaction mixture was quenched with 0.5% aqueous HCl, and the crude cyanohydrin was extracted twice with toluene. The combined organic layers were dried over Na_2SO_4 and concentrated in vacuo, producing a brown oil which was chromatographed on silica gel to give the pure cyanohydrin. The 1H NMR and IR spectra were found to be identical with those of an authentic sample. The optical yield of the resulting cyanohydrin ((*S*)-**2**: (*R*)-**2**) was determined by HPLC analysis (column, Sumipax OA-4100; eluent, hexane/1,2-dichloroethane/ethanol/ethyl acetate=800/200/10/1; flow rate, 1.0 mL min^{-1} ; detection, 254 nm light). The conversion of **1** was also determined by HPLC analysis (column, LiChrosorb SI-60; eluent, hexane/ethyl acetate/acetic acid=500/60/3; flow rate, 1.0 mL min^{-1} ; detection, 254 nm light).

References

- 1) a) T. Matsuo, T. Nishioka, M. Hirano, Y. Suzuki, K. Tsushima, N. Itaya, and H. Yoshioka, *Pestic. Sci.*, **11**, 202 (1980). b) K. Aketa, N. Ohno, and H. Yoshioka, *Agric. Biol. Chem.*, **42**, 895 (1978).
- 2) a) J. Oku and S. Inoue, *J. Chem. Soc., Chem. Commun.*, **1981**, 229. b) J. D. Elliott, V. M. F. Choi, and W. S. Johnson, *J. Org. Chem.*, **48**, 2294 (1983). c) M. T. Reetz, F. Kunishi, and P. Heitman, *Tetrahedron Lett.*, **39**, 4721 (1986). d) Y. Kobayashi, S. Asada, I. Watanabe, H. Hayashi, Y. Motoo, and S. Inoue, *Bull. Chem. Soc. Jpn.*, **59**, 983 (1986). e) H. Minamikawa, S. Hayakawa, S. Yamada, N. Iwasawa, and K. Narasaka, *Bull. Chem. Soc. Jpn.*, **61**, 4379 (1988). f) M. Hayashi, T. Matsuda, and N. Oguni, *J. Chem. Soc., Chem. Commun.*, **1990**, 1364.
- 3) a) A. Mori, Y. Ikeda, K. Kinoshita, and S. Inoue, *Chem. Lett.*, **1989**, 2119. b) K. Tanaka, A. Mori, and S. Inoue, *J. Org. Chem.*, **55**, 181 (1990).
- 4) The 1H NMR and IR spectra were found to be identical with those of the authentic sample. The absolute configuration of the resulting cyanohydrins ((*S*)-**2** and (*R*)-**2**) has already been determined.^{1b,3b)}